Enrolment No.

## GUJARAT TECHNOLOGICAL UNIVERSITY

**B.E. Sem-III** (Biotechnology) **Examination December 2009** 

Subject code: 130404

Subject Name: Organ	ic Chemistry and Unit Processes
Date: 21 /12 / 2009	Time: 11.00 am - 1.30 pm

**Total Marks: 70** 

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- 1. Attempt all questions.
- 2. Make suitable assumptions wherever necessary.
- 3. Figures to the right indicate full marks. What are organic reactions? Classify organic reactions and **06 Q.1** explain substitution reaction along with examples. **(b)** Give functional derivatives of carboxylic acids with examples. (c) Give the metallurgy of iron? 04 (a) What is meant by unit process? What is its utility for Q.206 industrial purpose? Explain nitration. **(b)** Mention general steps involved in metallurgy with suitable examples. (c) Convert: 04 i) Toluene to benzyl chloride ii)Naphthalene to 1,8 – di nitro naphthalene OR Give physico-chemical applications of Uranium. 04 (c) Fill in the blanks: 04 Substitution reaction takes place in type of hydrocarbons. ii) Carbohydrates are \_\_\_\_\_ aldehydes and ketones.
  - iii) Glucose is type of sugar. iv) Transfer of electron results to \_\_\_\_\_
- Q.3How can the mechanism be studied for organic compounds? Explain in detail. **(b)** Explain hybridization taking place in methane and ammonia molecule.
  - (c) Mention principle involved in concentration techniques for metals.

## OR

- (a) What is hybridization? Mention types of hybridization with Q.306 suitable examples. Explain oxidation reaction with suitable examples. 04 (c) What is thermite process? What is its utility? 04
- **Q.4** What are carbohydrates? Classify giving suitable examples **06** and explain the structure of glucose in detail.
  - **(b)** How is aldo hexoses converted to keto hexoses? 04 (c) Explain Walden inversion with example. 04

Q.4	(a) (b)	formed when flux combines with gangue impurity in	06 04
	(c)	metallurgical processes. Explain amination by aminolysis.	04
Q.5	(a) (b)	What are carboxylic acids and give IUPAC names of straight chain saturated compounds containing carbon atom	06 04
	(c)	16 and 18. Explain halogenations on the basis of unit process with examples.	04
		OR	
Q.5	(a)	-	06
		i) Enantiomerism and distereomerism	
	<b>(b)</b>	ii) Recemic modification.  Montion physics showing property of steel with its	Ω4
	<b>(b)</b>	Mention physico – chemical property of steel with its application.	04
	(c)	Explain concept of Chirality.	04
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		Explain concept of Chirality.  ***********************************	