

LAB V-Fx-1.09 G-Scan-31  
1/05/09  
Con. 3473-09.

ME (M) Thermal sem II (K)

Cryogenics

BB-5769

(4 Hours)

[Total Marks : 100

- N.B. :** (1) Question No. 1 is compulsory.  
 (2) Answer any four questions out of remaining six questions.  
 (3) Answers to all parts of a question should be written together and one below the other.  
 (4) Assume any suitable data wherever required but justify the same.  
 (5) Use of T-s charts, h-c charts are permitted.

- ME (M) Thermal sem II Rev Cryogenics, 21/5/09  
MADAN
- 1 kg of air at 1 atmosphere pressure and 300K is to be liquefied. Find the amount of work input required if this has to be done using a reversed Carnot cycle. Why Carnot cycle is not the ideal cycle for liquefaction? 5
    - Derive an expression for ideal work of liquefaction and find the minimum work requirement to liquefy air that is at 1 atm and 300K. Comment on the pressure levels involved. 5
    - A simple Linde Hampson cycle operates between the pressure levels of 1 atm. and 100 atm., with isothermal compression at 300K. Find heat exchanger effectiveness when yield is 70% of the maximum. Show schematic diagram and trace the cycle on T-s chart and derive all equations used. 10
  - Explain the working of the pre cooled Linde Hampson cycle. Is there a restriction on increase of refrigerant flow rate? 6
    - An air liquefaction system operates on pre cooled L-H cycle with pre cooling of air at  $-40^{\circ}\text{C}$  by using an ammonia vapor compression system with specific power consumption of 8540 kJ/kWh. The pressure levels of the main system are 1 atm. and 200 atm. respectively. Temperature of isothermal compression is 305 K. Heat gain into storage vessel is 8.8 kJ/kg. Determine the flow rate of air to be compressed to obtain 50 kg/h of liquid air. Also determine power required for the same. 14
  - List various insulations used in cryogenic systems in order of increasing performance and describe them in detail. 10
    - With a neat sketch describe the elements of a Dewar vessel clearly outlining the function of each of the elements. 10
  - A mixture consisting of 73% nitrogen, 18% oxygen and 9% argon by volume is to be separated into their pure constituents. The work required is 125 kJ/kg mixture. Find figure of merit of the system. Derive any formula used. 10
    - A rectification column is to be used to separate a feed stream of saturated liquid of composition 79% nitrogen and 21% oxygen into two streams. The top product being saturated liquid containing 98% nitrogen and bottom product saturated liquid with 95% oxygen. Derive the equation of the operating lines of the column and hence explain the McCabe Thiele method to find the number of plates needed for this duty. 10

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M. E. (Thermal Engg) Sem IV Cryogenics

- 5. (a) Derive an expression for the thermodynamically ideal isobaric-source system and show that it reduces to the expression of the Carnot refrigerator when  $T_2/T_1$  approaches unity. 8
- (b) Describe working of Philips cryocooler stating the importance of regenerator effectiveness. 12
- 6. (a) Define Murphree efficiency and discuss the factors that lead to high values of Murphree efficiencies. 10
- (b) Describe temperature measurement at cryogenic temperature levels. 10
- 7. Write short notes on any four of the following :— 20
  - (a) Linde double column separation system
  - (b) Cryogenic applications in space technology
  - (c) Cryopumping
  - (d) Simon helium liquefaction system
  - (e) Ortho-para conversion of Hydrogen.

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